



COMMENTARY

Selective Pharmacological Inhibition of Distinct Nitric Oxide Synthase Isoforms

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ABSTRACT. Nitric oxide (NO) is produced in physiological and pathophysiological conditions by three distinct isoforms of NO synthase (NOS): endothelial NOS (ecNOS), inducible NOS (iNOS), and brain NOS (bNOS). Selective inhibition of iNOS may be beneficial in various forms of shock and inflammation, whereas inhibition of bNOS may protect against neuroinjury. This article surveys the enzymatic mechanism of NO production, lists the strategies and pharmacological tools for selective inhibition of distinct NOS isoforms, and considers the side-effects of the various approaches. Selective inhibition of NOS isoforms is achieved by: (a) targeting the differential co-factor (calmodulin or tetrahydrobiopterin) requirement of various NOS isoforms of NOS; (b) targeting the differential substrate requirements of cells expressing various isoforms of NOS (L-arginine uptake blockers or arginase); (c) the use of pharmacological agents that are selectively taken up by cells expressing various isoforms of NOS (7-nitroindazole); or (d) developing pharmacological NOS inhibitors with isoform specificity. The amino acid-based NOS inhibitor, N^G -nitro-L-arginine, shows a preference for ecNOS and bNOS over iNOS, whereas L- N^6 -(1-iminoethyl)lysine is selective for iNOS over bNOS. Certain non-amino acid-based small molecules, such as aminoguanidine and certain S-alkylated isothioureas, also express selectivity towards iNOS and have anti-inflammatory and anti-shock properties. 7-Nitroindazole, a bNOS-selective inhibitor, protects in central nervous system injury. Clearly, there are a number of distinct approaches that are worthy of further research efforts in order to achieve even more selective targeting of various NOS isoforms. *BIOCHEM PHARMACOL* 51;4:383–394, 1996.

KEY WORDS. nitric oxide; endothelium; macrophage; selectivity; L-arginine; N^G -nitro-L-arginine; 7-nitroindazole; isothioureas; amidines; aminoguanidine; inflammation; shock; vascular effects

ACTIONS AND BIOLOGICAL ROLES OF NO† *Physiological and Pathophysiological Roles of NO*

NO is a small membrane-permeable gas that serves as a mediator of many physiological events. It is produced by the oxidation of L-arginine by a family of isoenzymes—NOS [1–3]. NO produced by the ecNOS is involved in the regulation of blood pressure, organ blood flow distribution, and the inhibition of the adhesion and activation of platelets and polymorphonuclear granulocytes [4]. Inhibition of the actions of ecNOS by N^G -substituted analogues of L-arginine, such as L-NMA or L-NA, causes vasoconstriction, increases blood pressure, and reduces blood flow to organs. Dysfunction of NO formation in the vascular endothelium has been implicated in the pathophysiology of a number of vascular and non-vascular diseases [4, 5].

The NOS isoform present in the central and peripheral

nervous system (bNOS) is another constitutive, calcium/calmodulin-dependent enzyme [6]. NO derived from bNOS has important physiological roles as a neurotransmitter and as a potential mediator of the metabolism/blood flow coupling in the brain. Enhanced formation of NO (e.g. following activation of *N*-methyl-D-aspartate receptors or induction of iNOS in the microglia) plays a role in neuroinjury [7, 8], suggesting that inhibition of bNOS may be of therapeutic benefit in central nervous system injury.

A distinct isoform of NOS (iNOS) can be induced by pro-inflammatory agents such as endotoxin (bacterial lipopolysaccharide, LPS), interleukin-1 β , tumor necrosis factor- α and interferon- γ in a variety of cells, including macrophages and smooth muscle cells [9]. Enhanced formation of NO following the induction of iNOS has been implicated in the pathogenesis of a number of conditions, including various forms of circulatory shock [10–13] and inflammation [14, 15]. The pathophysiological importance of an enhanced formation of NO by iNOS suggests that inhibitors of iNOS have therapeutic potential, particularly those that do not interfere with the protective and physiological roles of ecNOS.

NOS: Catalytic Activity and Sites of Potential Inhibition

NOS produces NO by the oxidation of one of the guanidino nitrogens of arginine (or certain other substrates). The process

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† Abbreviations: BH₄, tetrahydrobiopterin; L-NA, N^G -nitro-L-arginine; L-NAME, N^G -nitro-L-arginine methyl ester; L-NMA, N^G -methyl-L-arginine; NO, nitric oxide; NOS, nitric oxide synthase(s); bNOS, brain nitric oxide synthase isoform; ecNOS, endothelial constitutive nitric oxide synthase isoform; iNOS, inducible nitric oxide synthase isoform; and ITU, isothiourea.

involves the oxidation of NADPH and the reduction of molecular oxygen. The transformation occurs at a catalytic site adjacent to a specific binding site for L-arginine. The catalytic site includes an iron-containing heme group, iron protoporphyrin IX, in which the ferric iron is coordinated to the tetrahentate porphyrin in a planar arrangement. The iron is also coordinated to the thiol of a cysteine residue (thought to be cysteine 184 in ecNOS and cysteine 415 in bNOS [16, 17]), thus linking the heme to the rest of the protein. The sixth coordination site is available for the binding and activation of molecular oxygen during the stepwise oxidation of L-arginine [2, 3].

Flavin-adenine dinucleotide and flavin mononucleotide are both co-factors and, by analogy with their function in P450 reductases, as thought to be involved in the transfer of electrons from NADPH to the catalytic center. It has been suggested that the flavoprotein inhibitor, diphenyleneiodonium, exerts its inhibition of NOS activity by its irreversible effects on the FAD binding site [18].

Cloning of the isoenzymes has revealed that NOS subunits are divided into a reductase domain and an oxygenase domain, the latter containing the catalytic center. These domains are linked by a sequence that can bind calmodulin. The binding of calmodulin appears to be essential for NOS activity and is thought to control the electron transfer from the flavins to the heme, possibly by causing a reorientation of the reductase and oxygenase domains to allow electron transfer between them. The binding of calmodulin to native (inactive) constitutive isoforms of NOS is promoted by calcium ions. Consequently, the activities of bNOS and ecNOS are regulated by the concentrations of intracellular calcium. In contrast, calmodulin is tightly bound as a prosthetic group to the inducible NOS holding it in an activated state that is independent of intracellular calcium.

The oxygen domain of NOS has a highly conserved 320-amino acid sequence that is likely to contain binding sites for BH₄ as well as heme and L-arginine, which are likely to be in close proximity [19]. The function of BH₄ is unclear, but it may act as an allosteric effector or stabilizer of the active structure of NOS (i.e. assembling dimeric forms of NOS in the case of bNOS and iNOS) [2, 3].

The synthesis of NO by NOS involves the binding of L-arginine so that the guanidino group is adjacent to the ferric iron of the heme prosthetic group. The iron is reduced by transfer of an electron from NADPH and binds molecular oxygen, which is then cleaved, with one oxygen atom released as water and the other incorporated onto one of the terminal guanidino nitrogens of arginine to yield hydroxyarginine. Subsequent activation of a further molecule of oxygen facilitates the further oxidation of hydroxyarginine to produce water, NO, and citrulline. The chemistry of the process remains to be clarified (see Refs. 2, 3, and 20–22 for details and discussion), but it is evident that the guanidino group of the substrate must be in close proximity to the (normally vacant) sixth coordination site of the heme iron where molecular oxygen binds and becomes activated for subsequent reaction with substrate. Coordination of a sixth ligand perturbs the arrangement of the iron within the porphyrin and results in a spectral shift that is dependent on the ligand incorporated [2, 3, 22]. Occupation of

this site by ligands other than oxygen would prevent the oxidation of L-arginine and is probably involved in the mode of action of NOS inhibition by small molecules such as carbon monoxide, NO (although thiols and amines are other potential sites), and many nitrogen-containing species. This mechanism on its own is unlikely to provide any discrimination between various NOS isoforms.

NOS isoforms have narrow specificity with respect to the substrates utilized to produce NO. Besides the natural substrate, L-arginine, the only other compounds reported or proposed as substrates for NOS are N α -benzoyl-L-arginine, L-homoarginine, ϵ -guanidinocaproic acid, canavanine, L-N^G-methoxyarginine and L-N^G-hydroxyarginine [23, 24]. L-NMA, a widely used inhibitor of NOS (see below), is also capable of NOS-dependent generation of NO [25], although the proposed mechanisms suggest either L-arginine or hydroxyarginine as intermediates. The D-isomers of the above-listed compounds are not substrates, nor are they inhibitors. Interestingly, in the absence of substrate, NOS has been shown to produce superoxide anion and hydrogen peroxide [26].

iNOS utilizes large amounts of L-arginine, and this may result in depletion of intracellular L-arginine pools and/or an increased uptake of L-arginine from extracellular sources. Induction of iNOS is also associated with the induction of arginosuccinate synthetase, which, by stimulating the "re-cycling" of L-citrulline to L-arginine, may support iNOS with its substrate from endogenous sources [27]. In some cells, the same stimuli that induce iNOS also enhance the membrane transport system for L-arginine [28]; prolonged activation of iNOS may require the replacement of intracellular L-arginine from extracellular sources [29]. Some of the NOS inhibitors (e.g. L-NMA) also inhibit the transporter system, whereas others (e.g. L-NA) do not affect it [30]. The transport systems can also be inhibited in a competitive fashion by cationic amino acids such as L-lysine and L-ornithine [29]. The extensive extracellular L-arginine requirement of cells expressing iNOS may offer a target for pharmacological intervention: extracellular arginase [31] or inhibition of L-arginine transport with specific agents such as the tetravalent guanylhydrazone CNI-1493 [32] has been shown to provide therapeutic benefit in experimental models of septic shock.

Differential co-factor requirements of various NOS isoforms offer a way for selectively targeting NOS isoforms. For instance, the calmodulin-dependent NOS isoforms (ecNOS and bNOS), as with most calmodulin-dependent enzymes, can be inhibited by calmodulin inhibitors, such as calmidazolium [33]. iNOS, which contains calmodulin in a tightly bound form, is not inhibited by these agents. While antagonism of calmodulin can provide a basis for selectivity, the existence of many other calmodulin-dependent biological processes makes the usefulness of this approach questionable.

BH₄ has been suggested to be a critical co-factor for iNOS. Inhibition of BH₄ production by 2,4-diamino-6-hydroxypyrimidine, a selective inhibitor of GTP cyclohydrolase I, the rate-limiting enzyme in the production of BH₄, inhibits NO production by iNOS [34, 35]. In addition, it has been shown that N-acetylserotonin, phenprocoumon and dicumarol, which inhibit the synthesis of BH₄ by blocking sepiapterin

reductase, also reduce cytokine-induced NO production [35, 36]. These data suggest that inhibitors of BH₄ biosynthesis may be useful in preventing NO formation by iNOS, especially since phenprocoumon and dicumarol, while inhibiting iNOS activity, do not block NO production by eNOS [36]. In apparent agreement with the *in vitro* results, *N*-acetylserotonin pretreatment has been shown to reduce endotoxin-induced hypotension in the rat [37]. However, recent data indicate that *N*-acetylserotonin also inhibits the induction of iNOS [38], thus making the prior experimental results based on pharmacological actions of *N*-acetylserotonin difficult to interpret. The use of BH₄ synthesis inhibitors may also be hampered by the fact that BH₄ is a co-factor for tyrosine hydroxylase and tryptophan hydroxylase (among other enzymes) and so inhibition of BH₄ production may lead to reduced biosynthesis of dopamine, epinephrine, norepinephrine, and serotonin [39–41].

It has been suggested that modulation of NOS activity can be achieved by selective phosphorylation of NOS as illustrated by the neuroprotective effect of the immunosuppressant drug FK506. This drug appears to inactivate iNOS by inhibiting its dephosphorylation [42]. Although similar mechanisms have not been reported for eNOS or iNOS, the isoform-selectivity of the above process needs to be further explored.

STRUCTURE AND BIOLOGICAL ACTIONS OF L-ARGININE-BASED INHIBITORS OF NO SYNTHASES

Many analogues of L-arginine are inhibitors of NOS, in particular those substituted at one or both of the terminal guan-

dino (amidino or ω-) nitrogens [22, 43] (Fig. 1). On the whole, the L-isomers are far more effective inhibitors of NOS than the corresponding D-isomers, which are not generally inhibitors, although one exception is known: both D-NA and L-NA inhibit endothelium-dependent relaxations in rat aortic rings [44]. Compounds substituted on the propylene chain are generally neither inhibitors nor substrates. Similarly, removal of the α-amino or α-carboxyl groups reduces recognition by the arginine binding site. For instance, 5-guanidino pentanoic acid (argininic acid) and 4-aminobutylguanidine (agmatine) are neither substrates nor noteworthy inhibitors of either bNOS or iNOS [24, 25].

Even esterification of the carboxyl group can reduce inhibitory potency markedly. L-NAME, often regarded as equivalent to L-NA, is, in fact, over an order of magnitude less potent towards iNOS than L-NA [45]. In principle, the fact that esterase activity is needed for L-NAME to fully exhibit its inhibitory action may offer the opportunity for selective targeting of NOS in various tissues (with specific esterase activities) using appropriate esters of NOS inhibitors.

All of the NOS isoforms can be inhibited (to a variable degree) with N^G-substituted L-arginine analogues, e.g. L-NMA. Some analogues of L-arginine do exhibit some isoform selectivity, mostly towards the constitutive isoforms. N^G-Cyclopropyl-L-arginine shows a preference for bNOS over iNOS *in vitro* [46], whereas L-NA and L-NAME (after hydrolysis) show selectivity toward eNOS. L-NMA and N^G-amino-L-arginine show no marked preference for either isoform [46–50]. The mechanisms of inhibition of NOS activity by these compounds vary, but all involve occupation of the substrate bind-

	R ₁	R ₂	n
L-arginine	NH ₂	NH	3
L-NMA	NH ₂	NCH ₃	3
L-NA	NH ₂	NNO ₂	3
L-NAME*	NH ₂	NNO ₂	3
L-amino-arginine	NH ₂	NNH ₂	3
L-cyclopropyl-arginine	NH ₂	NC ₃ H ₅	3
L-allyl-arginine	NH ₂	NCH ₂ CH ₃	3
L-iminoethyl-ornithine	CH ₃	NH	3
L-N ⁶ -iminoethyl-lysine	CH ₃	NH	4
L-homoarginine	NH ₂	NH	4
L-citrulline	NH ₂	O	3
L-thiocitrulline	NH ₂	S	3
S-methyl-thiocitrulline	SCH	NH	3
S-ethyl-thiocitrulline	SCH ₂ CH ₃	NH	3
L-hydroxyarginine	NH ₂	NOH	3

* in L-NAME, the acid function (- COO⁻) becomes (- CO - O CH₃)

FIG. 1. Structures of L-arginine, L-citrulline, L-hydroxyarginine, L-homoarginine and various amino acid-based NOS inhibitors.

ing site. Simplistically, this excludes L-arginine and hence prevents its metabolism to NO. Indeed, many L-arginine-based inhibitors (cyclopropyl-L-arginine, L-NA) are competitive inhibitors of NOS in that their effects can be reversed by L-arginine. However, their influence on electron transfer and reduction of the heme iron in the NOS molecule may differ. L-NAME appears to inhibit NADPH consumption by interrupting electron flux immediately prior to reduction of the heme iron, whereas L-NMA does not influence the reduction of the heme iron, but uncouples NADPH oxidation from catalytic activity [51]. Some compounds such, as *N*-iminoethyl-L-ornithine [52], show inhibition that cannot be reversed by L-arginine (or other factors). Such irreversible inhibition can be associated with alterations of inhibitor binding to reduce the rate of dissociation of the inhibitor/NOS complex or chemical modification of the inhibitor and/or enzyme. For instance, exposure of iNOS to L-NMA results in a time-dependent irreversible inactivation of the enzyme, which is preceded by an NADPH-dependent hydroxylation of the inhibitor to *N*-hydroxy-*N*-methyl-L-arginine [20]. In addition, L-NMA may cause a significant loss of heme [53] from the enzyme. Similarly, the irreversible component of the inhibition of NOS by *N*^G-allyl-L-arginine probably involves generation of an intermediate capable of covalent binding to the enzyme. Interestingly, irreversible inhibition of bNOS by L-NA has also been reported [54], in contrast to the reversible nature of its inhibition of iNOS. This illustrates differences in the binding characteristics of NOS isoforms that offer a potential rational basis for drug development.

Dimethylarginines, such as symmetric (*N,N'*-dimethylarginine) and asymmetric dimethylarginine (*N,N*-dimethylarginine) are endogenously produced inhibitors of NOS, but do not show any isoform selectivity [23]. However, the plasma levels of these compounds are known to increase in chronic renal failure [55, 56] and in hypercholesterolemia [57].

Despite their lack of marked isoform selectivity and certain non-specific effects (see below), L-arginine-based inhibitors have been used widely in the last 5 years in order to elucidate the pathophysiological importance of NO and have demonstrated beneficial effects in various forms of shock, inflammation, and central nervous system disorders [1, 2, 5, 7, 11–14, 42, 43, 58–66]. It appears that in shock and inflammation, NO derived from iNOS is responsible for most of the pathophysiological actions, whereas, in many cases, ecNOS has beneficial and protective roles and so its inhibition is not advantageous. Excessive inhibition of ecNOS may mask the beneficial effects of inhibiting iNOS, as in endotoxin-induced intestinal injury [64] or liver failure [65–67]. It follows that L-arginine-based inhibitors of NOS may not be ideal when targeting a particular isoform of NOS (especially iNOS) due to their limited selectivity and the relative low potency. The ready *in vitro* reversibility of their effects by L-arginine, together with the fact that millimolar concentrations of L-arginine are present in the plasma, may explain the need for large doses of these agents to achieve biological effects.

The requirement for high doses of inhibitors not only exaggerates their effects on other isoforms of NOS, but also

reveals an increasing number of non-specific actions of L-arginine-based inhibitors. Some of these effects impinge on the arginine-NO system: L-NMA, but not L-NA, affects arginine transport into the cell (a rate-limiting step in NO production by iNOS) [30]. L-NA has been shown to inhibit arginase activity (which catalyzes the conversion of L-arginine to L-ornithine), while L-NMA, L-NAME and aminoguanidine do not [68]. These effects are properties of a particular inhibitor, rather than general properties of all L-arginine-based inhibitors. L-NMA can act as a substrate for NO production by NOS in the rat aorta and rabbit pulmonary artery [25]. In isolated rat aortic rings, L-NMA (as well as L-NA) inhibits the “agonist-induced” endothelium-dependent (ecNOS-mediated) relaxations elicited by acetylcholine, but (unlike L-NA) does not augment the phenylephrine-induced contractions in the same vessel [69], even though this augmentation is generally considered as an indicator of “basal” release of NO by ecNOS. Although there is some speculation about different types of endothelial NOS or changes induced in the enzyme that allow utilization of L-NMA as a substrate [25, 69], there is no definitive explanation for these findings. Another commonly used agent, L-NAME, requires hydrolysis to become effective against iNOS. This probably occurs rapidly *in vivo*, but can lead to possible complications *in vitro*, where short incubation times, leading to partial hydrolysis, are used. Similar considerations apply when using other esters and amides of arginine analogues (e.g. L-NA-*p*-nitroanilide; [45]). L-NAME and other esters of L-arginine, but not other NOS inhibitors tested (such as L-NMA), have been reported to act as muscarinic receptor antagonists [70].

More general, non-NOS-specific properties of arginine analogues may include their inhibition of the activity of iron-containing enzymes such as catalase [71] and effects on iron-containing systems, in general, as illustrated by the attenuation by NOS inhibitors of the inhibitory effect of ferrous iron on the reduction of cytochrome *c* [72].

In addition to these non-specific effects of L-arginine-based NOS inhibitors *in vitro*, unexpected side-effects of NOS inhibitors can also be observed *in vivo* in some cases. For instance, L-aminoarginine has been reported to cause epileptiform seizures in dogs [73], an effect that may be related either to effects of NO blockade in the central nervous system, or, alternatively, to a non-specific effect of the drug itself.

OTHER AMINO ACID-BASED INHIBITORS OF NO SYNTHASES

Recent work has demonstrated that derivatives of L-citrulline and L-lysine can be potent inhibitors of NOS [74–76], some of them with isoform selectivity. L-Thiocitrulline (Fig. 1) is an extremely potent inhibitor of all NOS isoforms, which binds to the heme moiety of the NOS molecule, thereby reducing its redox potential [51]; its effects are reversed dose dependently by L-arginine [77]. L-Thiocitrulline has no particular isoform selectivity [77] and elicits similar potent pressor effects in both normal and endotoxemic rats [77], suggesting that it can be only of limited therapeutic use for targeting iNOS. In contrast, S-ethyl- and S-methyl-thiocitrulline (both of which can be

considered as isothioureas, see below and Fig. 1) are more potent inhibitors of bNOS than of ecNOS or iNOS [74], and thus may be of therapeutic potential in various neurodegenerative diseases that are associated with NO overproduction by bNOS. However, the therapeutic usefulness of S-ethyl- and S-methyl-thiocitrulline may be limited due to its poor uptake into cells [74]. Their effects are arginine-reversible and, like L-NA (but unlike L-NMA), they inhibit the NADPH oxidase activity of bNOS [74].

Recently, the Monsanto Co. has published on the development and some biological actions of the novel NOS inhibitor L-N⁶-(1-iminoethyl)lysine, which exhibits a 30-fold selectivity for iNOS over bNOS [76, 78]. Although data regarding the selectivity of the agent on iNOS over ecNOS have not been disclosed, preliminary data indicate that the compound exerts potent anti-inflammatory actions in a rat model of adjuvant arthritis [78].

NON-AMINO ACID-BASED INHIBITORS OF NO SYNTHASES

Arginine and its analogues can be regarded as derivatives of guanidine, so that the boundary between an arginine analogue and a non-arginine-based compound is vague and somewhat arbitrary. For the purposes of this review, we will consider compounds without both α -amino- and α -carboxylic acid groups as non-arginine-based compounds as they no longer possess the amino acid group, which has its own requirements for binding to a portion of the substrate binding site of NOS.

Non-arginine-based inhibitors of NOS represent probes to determine the structure of the L-arginine binding site/catalytic site of iNOS without constraints imposed by the specific stereochemistry of the binding of an amino acid group. Hence, they may prove to be useful tools in the design of isoform-specific inhibitors. Furthermore, demonstration of anti-NOS activity may now provide explanation for the mechanism of previously known actions of various biologically active compounds (see below).

Despite a requirement for the conserved stereochemistry of arginine analogues, a variety of compounds that are not amino acids are inhibitors of NOS activity. Examples include imidazoles [79, 80], 7-nitroindazoles [81], guanidines [82], iminobiotin [83], ebselen and derivatives [84], α -guanidinoglutaric acid [85], tannin [86], 3-amino-1,2,4-triazole [87], phencyclidine [88], and methylene blue [89]. Among this large and continually increasing list of inhibitors, only a few groups of compounds, namely aminoguanidines, isothioureas, and certain imidazoles and indazoles, have been reported to exhibit isoform selectivity and will be discussed below.

Guanidines (Fig. 2b) have long been known to cause pressor responses in various animals, and their similarity to L-arginine (a substituted guanidine) led to their investigation as inhibitors of NOS. Guanidine itself is not an inhibitor, but methylguanidine (an endogenous substance) and ethylguanidine do inhibit NOS, although they are weaker than L-NMA (authors' unpublished observations). 1,1-Dimethylguanidine is somewhat more potent against iNOS (induced in insulinoma cells) than methylguanidine, but is still weaker than L-NMA. How-

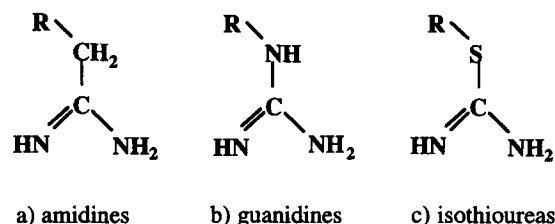


FIG. 2. General structures of amidine derivatives.

ever, it appears to be equivalent to L-NMA in its effect on ecNOS in mesenteric artery preparations and causes rapid increases in MAP *in vivo*. Arcaine, containing two guanidine groups (1,4 diguanidinobutane), shows moderate inhibition of NOS [24].

Other alkyl-substituted guanidines show weak or negligible effects on NOS. On the other hand, *aminoguanidine* (Fig. 2b, R = -NH₂) has received much attention as an inhibitor of NOS due to the early recognition of its selectivity towards iNOS [90], its low acute toxicity, and its potential clinical usefulness. There are now numerous reports on the beneficial effects of aminoguanidine in various experimental models of inflammation and shock. In most *in vitro* systems, aminoguanidine is nearly equipotent with L-NMA in its ability to inhibit NO production by immunostimulated macrophages. On the other hand, aminoguanidine is less potent than L-NMA on ecNOS and bNOS activities in cell homogenates [65, 91] and on ecNOS *in vivo* (as demonstrated by its limited pressor effects in anesthetized rats). However, it has been noted that when aminoguanidine is incubated with cells or tissues containing NOS or with semi-purified NOS, the length of the incubation affects the apparent potency: short incubation times result in a reduced inhibitor action [91, 92]. Such a difference is not found for other inhibitors such as S-methyl-isothiourea (see below) or L-NMA. Moreover, in certain experimental systems the selectivity of aminoguanidine towards iNOS appears to be rather limited, for example, when comparing its effects on ecNOS and iNOS from intestinal homogenates [93]. In *in vivo* experiments, doses of aminoguanidine required to elicit a considerable inhibition of iNOS (and thus, elicit beneficial biological effects) are rather high: 15–45 mg/kg, i.v., in anesthetized rats or up to 400 mg/kg/day in mice [91, 94, 95].

Aminoguanidine has been studied extensively in diabetes where it appears to exert NO-dependent and NO-independent beneficial effects. Aminoguanidine prevents the interleukin-1-induced decrease in glucose secretion in islets of Langerhans and reduces the glucose-induced vascular dysfunction in rats [90, 96]. Its effects in inhibiting diabetic vascular dysfunction also appear to be NO-mediated [97], whereas the NO-dependent [61, 62] development of diabetes in a low-dose streptozotocin model may [94] or may not [98] be prevented by aminoguanidine. Aminoguanidine has been shown to restore contractile responses in vascular tissues of animals treated with endotoxin *in vivo* and *in vitro* (an iNOS-dependent phenomenon) [90–92, 99]. In addition, aminoguanidine improves the survival rate in mice treated with endo-

toxin [91] to an extent comparable with the improvement in survival seen in the iNOS knock-out transgenic mice treated with endotoxin [66].

Other effects of aminoguanidine include anti-inflammatory actions: it inhibits the progression of experimental autoimmune encephalomyelitis [95] and the cardiovascular consequences of experimental pancreatitis [100], as well as prevention of the endotoxin-induced acute pulmonary extravasation [101]. All these biological effects of aminoguanidine have been tentatively linked to its inhibitory effect on iNOS. The clear evaluation of the *in vivo* effects of aminoguanidine as an inhibitor of iNOS, however, is hampered by the fact that it has numerous other effects. These include inhibition of histamine metabolism [102], inhibition of polyamine catabolism [103], and inhibition of catalase [104] and other copper- or iron-containing enzymes. Other beneficial, but NO-independent, effects of aminoguanidine are its reduction of end product glycosylation in diabetes and its inhibitory effect on oxidative modification of low density lipoprotein [105]. These effects may be consequences of the reactivity of aminoguanidine in providing an alternative substrate for glycosylation and, in the latter case, by reacting with aldehydes formed during lipid peroxidation. Aminoguanidine also potentiates the effect on angiotensin-induced prostacyclin production in rat lungs, an effect not seen with L-NA [106]. Apparently, aminoguanidine can be tolerated in high doses in experimental animals, although chronic administration may cause side-effects, as its hydrolysis can give rise to semicarbazide and hydrazine, both toxic products.

Certain derivatives of aminoguanidine, such as *N,N'*-diaminoguanidine, although exhibiting lower potency than aminoguanidine, appear to be selective for iNOS [82]. The mechanism of inhibition of NOS by aminoguanidines (and possibly guanidines) if thought to be due to their binding as ligands to the heme iron at the catalytic site as aminoguanidine deactivates other iron- or copper-containing enzymes in this manner. Competitive binding with L-arginine is probably involved as L-arginine can attenuate the pressor responses to some guanidines [82]. However, this explanation does not address the apparent selectivity of some potential ligands and not others. Recent data show that the inhibition of NOS by aminoguanidine becomes greater with increasing incubation time [91, 92, 107], indicating that aminoguanidine is a mechanism-based inhibitor [107]. Inactivation of various catalytic functions of NOS by aminoguanidine (NADPH oxidase and citrulline formation but not the cytochrome c reductase activity) is dependent on an active NOS and the appropriate NOS co-factors [107].

Several compounds reported individually to inhibit NOS also contain the guanidino group. Inhibition of NOS activity by α -guanidinoglutaric acid [85] and by 3-amino-1,2,4-triazole [87] has been reported, but only for one isoform each (bNOS and iNOS, respectively). Iminobiotin [83], which is structurally related to mercaptoalkylguanidines (below), shows no discrimination between iNOS and bNOS.

The most potent guanidino inhibitors of iNOS reported to date are *mercaptoalkyl-guanidines*, in particular mercaptoethyl-

and mercaptopropyl-guanidines (Fig. 2c, $R = -CH_2CH_2SH$ and $-CH_2CH_2CH_2SH$, respectively) [108]. They are far more effective than L-NMA at inhibiting iNOS in whole cells, but mercaptoethylguanidine is much less effective against ecNOS *in vitro* and causes only slight pressor responses in anesthetized rats [107]. Mercaptoethyl-guanidine (and possibly the isothiourea, 2-aminothiazoline) is probably the active form of aminoethyl-isothiourea (Fig. 2b, $R = -CH_2CH_2NH_2$)*, a recently identified inhibitor of NOS [109, 110]. Thus, the beneficial effects of aminoethyl-isothiourea on iNOS-dependent pathophysiological processes in endotoxemia (reversal of hypotension* and protection against endotoxin-induced liver damage [111]) are likely to be due to mercaptoethyl-guanidine. Delineation of the actions of mercaptoalkyl-guanidines and related compounds are hampered by the ease of their oxidation to the disulfide in aqueous solution, a process that occurs *in vivo* [112]. It is worthy of note that mercaptoalkyl-guanidines, including mercaptoethyl-guanidine, have been described as radioprotective agents [112, 113], in view of the recent report that inhibitors of NOS *in vivo* also protect against exposure to radiation [114]. Substituents on the sulfur, as in *S*-methyl- or *S*-ethyl-mercaptoalkyl-guanidines, appear to reduce the potency of NOS inhibition, but maintain the selectivity towards iNOS as compared with ecNOS (authors' unpublished observations). Such compounds may have the advantage that they do not dimerize. Little is known, as yet, about the nature of the inhibition of NOS by mercaptoalkyl-guanidines, other than it is of a competitive nature. By analogy with other guanidines, binding to the heme of NOS is likely either through the guanidino group or the sulfur. As with other inhibitors that contain sulfur or selenium (e.g. carboxyethylselen [84]), there is always the possibility of interaction with the cysteine residue that is normally coordinated to the heme iron.

Guanidines, in general, exhibit numerous pharmacological properties that are independent of their inhibition of NOS, including inhibition of the membrane sodium/potassium ATPase [115]. They also have diverse vascular effects. While guanidine (which is not an inhibitor of NOS) and its methylated derivatives increase blood pressure, ethyl-, butyl-, and hexyl-guanidines cause a temporary hypotension before the pressor response [116–118]. Previous observations on vascular effects suggest that guanidino compounds with bulky substituents possess a smooth muscle relaxing action [116–118], whereas smaller guanidines may have pressor effects [116–118] that may be, at least partially, related to inhibition of NOS. *In vivo* administration of guanidines can result in tachyphylaxis: pressor responses to successive doses of the same guanidine diminish rapidly [116–118]. The depressor response, however, is unaffected [116–118]. The above factors should be born in mind when interpreting the cardiovascular effects of guanidines.

S-Substituted *isothioureas* (ITUs, Fig. 2c) have been described recently as very potent inhibitors of NOS by two independent groups [65, 109, 110]. Not only are some of these

* Southan GJ, Zingarelli B, O'Connor MP, Salzman AL and Szabo C, Manuscript submitted for publication.

compounds over 10–20 times more potent than L-NMA against iNOS, ecNOS, and nNOS, but they possess variable selectivity towards the various isoforms. S-Ethyl-ITU (Fig. 2c, $R = -CH_2CH_3$) and S-isopropyl-ITU appear to be the most potent members of the group of the straight chain ITUs. Extension of the side chain (R , Fig. 2c) decreases the potency (n -propyl $>$ t -butyl $>$ n -butyl, etc.) [110]. S-Ethyl- and S-isopropyl-ITU do not show any preference for a particular isoform in rodent models [110], but isopropyl-ITU shows some selectivity towards iNOS in human systems [109]. However, S-methyl-ITU is more potent on iNOS than on ecNOS in rodents [65]. Substitution of either or both of the amidine nitrogens greatly reduces the potency of a particular S-substituted ITU, as exemplified by S-methyl ITUs [110]. The exception to this is the cyclic compound, 2-aminothiazoline (where the sulfur is attached via a 2-carbon chain to one of the nitrogens) and its methylated derivatives, which are still as potent as L-NMA, but have no isoform selectivity. A further group of very potent ITUs are of the bis-ITU type where the sulfur atoms of two isothiourea units are linked by a carbon chain sometimes containing unsaturated (hetero)-cyclic rings. Some of these bis-ITUs show marked selectivity for human iNOS when compared with human ecNOS and bNOS preparations [109]. Unfortunately, poor cellular uptake and direct cytotoxic effects will probably limit the therapeutic usefulness of this “bis” class of compounds [109]. Isothioureas have been demonstrated to perturb the heme spectra of NOS, suggesting binding close to, or interaction with, the heme center [109].

So far, there are a limited number of published reports on *in vivo* experiments investigating the effects of ITUs in various models associated with NO overproduction. In a rat model of endotoxin shock, S-methyl-ITU caused a restoration of the blood pressure and pressor (vasoconstrictor) responsiveness. Its pressor effect (similar to that of aminoguanidine, but unlike L-NMA or L-NA) is larger in endotoxin-treated rats than in control animals [65]. Moreover, S-methyl-ITU reduces the endotoxin-induced increase in the plasma markers of hepatic injury [65], reduces endotoxin-induced pulmonary leak in rats [101], and improves 24-hr survival in endotoxin-treated mice [65]. Similarly, S-methyl-ITU has beneficial effects in a gram-positive model of sepsis in rats [119]. In a hyperdynamic model of endotoxin shock in the sheep, S-ethyl-ITU has been shown to cause a marked and sustained increase in blood pressure [120]. S-Methyl-ITU [61] and S-aminoethyl-ITU are also effective in preventing endotoxin-induced liver failure [111], although the effects of the latter drug are presumably due to its metabolite, mercaptoethylguanidine (as discussed above). Interestingly, there is an early report demonstrating a marked pressor effect of S-methyl-ITU in a patient with spinal anesthesia [121]. This pressor effect is likely to represent the consequence of the inhibitory effect of the drug on ecNOS, rather than iNOS.

S-Methyl-ITU or other derivatives may cause tachyphylaxis in as much as pressor responses to subsequent injections of the agent or of other ITUs are attenuated in anesthetized animals. Indeed, this phenomenon has been noted for related pressor agents (isoureas, amidines) [116, 117, 122]. Thus, it appears

that ITUs, along with other derivatives of amidines, possess additional properties that affect the cardiovascular system.

Guanidines and isothioureas can be classified as derivatives of amidines (Fig. 2), of which formamidine (Fig. 2b, $-SR=H$) is the lowest member of the homologous series. Neither formamidine nor acetamidine (Fig. 2a, $R=H$) inhibits iNOS, but propionamidine (Fig. 2a, $R = -CH_3$) and longer chained amidines do [123]. Indeed, the cyclic amidine, 2-iminopiperidine, is a more potent inhibitor of iNOS than is L-NMA. Although some of these agents may have some preference for iNOS over ecNOS when compared with L-NMA, they are also powerful pressor agents in the rat, some disproportionately so when compared with their ability to inhibit NOS (e.g. butyramidine). Some amidines that are weak inhibitors of NOS are also potent pressor agents [116, 117], suggesting that these small amidines may exert their pressor actions through more than one mechanism. Long chain amidines (and long chain guanidines), on the other hand, have been observed to exhibit depressor actions. Of potential relevance to their NOS inhibitory actions is the ability of some derivatives of amidines to blunt the pressor effects of subsequent injections of the same or different classes of amidine derivatives (including guanidines)—a cross-species tachyphylaxis [122].

From the group of imidazoles and indazoles, 7-nitroindazole has received the most attention, as a selective inhibitor of bNOS. In contrast to other bNOS-selective agents, such as S-methyl- and S-ethyl-thiocitrulline or L-NA, the basis of selectivity appears to lie in the differential uptake of the inhibitor into endothelial cells versus neurons: 7-nitroindazole has minimal pressor effects *in vivo*, but is an extremely potent inhibitor of NOS activity in homogenates of endothelial cells [80, 81, 124]. The inhibition of NOS by 7-nitroindazole may involve multiple mechanisms: it binds to the heme group of NOS in an L-arginine and/or BH_4 reversible fashion, depending on the isoform of NOS studied [80], and so appears to affect both the pteridine and arginine binding sites of NOS [125]. The first striking biological action of 7-nitroindazole was its potent antinociceptive effect, which is related to inhibition of bNOS. Similar antinociceptive effects have been observed with L-NA but not L-NMA. There are characteristic changes in regional cerebral blood flow in animals treated with 7-nitroindazole that are distinct from the effect of inhibition of ecNOS activity [126]. Recently, 7-nitroindazole has been used successfully against bNOS to protect against neurotoxicity elicited either by middle cerebral artery occlusion [127] or by the administration of 1-methyl-4-phenylpyridinium (MTPT) [128] in murine models. 7-Nitroindazole also exhibits NO-independent vascular actions such as endothelium-independent relaxant effect of various arterial smooth muscles [129]. 7-Nitroindazole reduces neurogenic edema formation, presumably by blocking bNOS present in the peripheral nerves [130].

Of the structurally related imidazoles, imidazole itself and its phenyl-substituted derivatives have been shown to inhibit NOS activity [79, 80], although only 1-phenylimidazole shows any appreciable selectivity, with a K_i for macrophage iNOS fifty times lower than for bNOS or ecNOS. As with other imidazoles, 1-phenylimidazole appears to exert its inhibitory

effects by virtue of its ability to bind heme and prevent the binding and activation of oxygen. However, its binding characteristics vary with isoform, but it is competitive with BH₄ and/or L-arginine.

It is likely that many of these small nitrogen-containing molecules can interact with the heme of NOS. Imidazoles inhibit the activity of various heme-containing proteins by binding to the heme group. Aminoguanidine chelates copper and iron ions and inhibits the actions of many enzymes that contain them. However, compounds that chelate iron ions do not necessarily affect NOS activity (amidoximes, for example). In addition there must be steric/structural constraints in the close proximity to the heme. Indeed, the heme group may be regarded as a portion of the inhibitor binding site.

SUMMARY AND FUTURE DIRECTIONS

Clearly, the various pathophysiological conditions associated with alterations in the body's NO homeostasis make NOS an attractive target for drug development. However, the complexity of the actions of NO and the various isoforms of NOS makes it necessary to develop isoform-selective, or, ideally, even cell-selective inhibitors of NOS. Specific targeting of NOS activity may be achievable by: (1) targeting the specific co-factor requirement of the various isoforms (calmodulin, BH₄, etc.); (2) targeting the L-arginine uptake systems or the availability of L-arginine (in the case of iNOS); (3) exploiting the differential uptake and/or metabolic characteristics of cells that express NOS (e.g. 7-nitroindazole); or (4) developing agents that specifically suppress the catalytic activity of a selected isoform of NOS. In the past, the development of substrate antagonists as isoform selective inhibitors of NOS has focused on: (a) L-arginine-based inhibitors; (b) non-arginine-based, amino-acid derivative inhibitors; and more recently (c) small, non-amino acid-based inhibitors. All three approaches have been successful inasmuch as approach (a) yielded bNOS selective inhibitors such as L-NA and cyclopropyl-L-arginine, approach (b) yielded the iNOS selective L-N⁶-(1-iminoethyl)lysine, and approach (c) resulted in the discovery of the iNOS selective guanidines (aminoguanidine and mercaptoalkylguanidines) and various isothiourea derivatives.

Approach (c) may have advantages in that non-arginine compounds may yield more information about the catalytic site of NOS: the stringent requirements for the binding of the amino acid group of an arginine analogue restricts the freedom of the (modified) guanidino group to arrange itself within the guanidine binding pocket of the substrate binding site. Small non-amino acid molecules, however, could arrange themselves within the guanidine/catalytic pocket independently of any such constraints. The point can be illustrated by the example of S-methisothiurea, a potent and somewhat selective inhibitor of iNOS: when a 2-aminopentanoic acid side chain is attached to one of the nitrogens (to give S-methyl-L-thiocitrulline), the selectivity is altered in favor of bNOS. When the side chain is attached to the sulfur in place of the methyl

group to make an arginine analogue (that is, the ϵ -nitrogen of arginine is replaced by a sulfur), the potency is substantially reduced.

Future Trends

From recent reports it seems that many groups believe that the achievement of pharmacological inhibitors of NOS with isoform selectivity is a realistic goal. Isoform selective NOS inhibitors may have therapeutic and commercial value in the therapy of various inflammatory and neurodegenerative disorders. It is generally assumed that the amino acid group of L-arginine analogues binds to a pocket within the substrate binding site that exhibits fairly stringent steric requirements. However, there is no evidence to suggest that this binding *per se* imparts any isoform-selectivity to any amino acid inhibitor of NOS. As it seems unnecessary to incorporate a group that is equally well recognized by all isoforms of NOS into the structure of a potential selective inhibitor, it is the opinion of the authors that the study of non-amino acid amidine derivatives will be at least as productive as the continued study of arginine analogues in the search for selective inhibitors of iNOS. The difference between the isoform preferences of aminoguanidine and its arginine derivative, N^G-amino-L-arginine, is one example to illustrate this point. Aminoguanidine also demonstrates the potential for more general actions of non-arginine-based compounds that are unrelated to their effects on NOS. However, as mentioned above, L-arginine-based inhibitors of NOS also have numerous side-effects. Such effects have to be considered when interpreting experimental data obtained with NOS inhibitors of the various structural classes.

The discovery of the inhibitory properties of certain isothioureas and nitroindazoles has raised interest in non-amino acid compounds. Most inhibitors of NOS thus far reported contain the amidine function ($-C(-NH_2)=NH$). This group encompasses arginines, guanidines, isothioureas, amidines, and even imidazoles. As the number of inhibitor types broadens, more will be learned about the structural features consistent with selective inhibition of NOS isoforms. Moreover, structural features deemed significant in one class of compounds may be subsequently tried in others. In view of the respective effects of amidines and the bis-isothiureas on iNOS, for example, it is likely that analogous bis-amidines will be synthesized and found to be inhibitors of NOS. Similarly, the significance of sulfur in inhibitors (as in isothiureas and mercaptoalkylguanidines) will no doubt be investigated, spawning compounds with both amidine and sulfur (or possibly selenium) containing functionalities.

Another advantage in using small molecules is that the synthetic chemistry involved is usually less tedious and may encourage a wider range of groups to develop their thoughts on NOS inhibitors. Hopefully, this review may do likewise.

NOTE ADDED IN PROOF

We note with interest that a recent paper presented at the 4th "Biology of Nitric Oxide" meeting describes certain cyclic amidines as selective inhibitors of iNOS, and certain sulfur-containing amidines as selective inhibitors of bNOS [131].

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